

The University of Jordan / Physics Department  
 Solutions for suggested Problems  
 Giancoli / seventh edition / chapter 31  
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$$Q38] \quad ED_{\alpha} = ED_{x\text{-rays}}$$

$$AD_{\alpha} RBE_{\alpha} = AD_x RBE_x$$

$$350 \times 20 = AD_x \times 1$$

$$\Rightarrow AD_{x\text{-ray}} = 7000 \text{ rad.}$$

$$Q40] \quad AD = 2.5 \text{ Gy} = 2.5 \times \frac{1 \text{ J}}{\text{kg}} = 2.5 \text{ J/kg.}$$

$$m = 65 \text{ kg} \Rightarrow$$

$$\text{Absorbed energy} = 2.5 \frac{\text{J}}{\text{kg}} \times 65 \text{ kg} = 162.5 \text{ J.}$$

$$Q41] E_p = 1.2 \text{ MeV} = 1.2 \times 10^6 \times 1.6 \times 10^{-19} \text{ J}$$

$$\text{tumor mass} = 0.20 \text{ kg.}$$

$$a) ED = AD \times RBE \Rightarrow 1.0 \text{ rem} = AD \times (1)$$

$$\Rightarrow AD = 1.0 \text{ Rad} (= 0.01 \text{ Gy}).$$

b) absorbed energy by tumor  $E_{\text{tumor}}$

$$E_{\text{tumor}} = M_{\text{tumor}} \times AD$$

$$= 0.2 \text{ kg} \times 0.01 \text{ J/kg}$$

$$= 0.002 \text{ J.}$$

$\Rightarrow$  number of absorbed protons is  $n$

$$n = \frac{E_{\text{tumor}}}{E_p} = \frac{0.002 \text{ J}}{1.2 \times 10^6 \times 1.6 \times 10^{-19} \text{ J/proton}}$$

$$\therefore n \approx 1.04 \times 10^{10} \text{ protons.}$$

$$\text{Q44]} \quad A = 1.6 \text{ mCi}$$
$$= 1.6 \times 10^{-3} \times 3.7 \times 10^{10} \text{ Bq}$$

$$= 5.92 \times 10^7 \text{ Bq} \quad (1 \text{ Bq} = 1 \text{ decay/s})$$

Need to administer 32 Gy to a tumor.

$$1 \text{ mCi} \rightarrow 10 \text{ mGy/min}$$

$$1.6 \text{ mCi} \rightarrow X$$

$$\therefore X = \frac{1.6 \text{ mCi}}{1 \text{ mCi}} \times 10 \text{ mGy/min}$$

$$= 16 \text{ mGy/min}$$

$\Rightarrow$  a 1.6 mCi delivers 16 mGy/min

required time  $t$  is given by

$$t = \frac{32 \text{ Gy}}{16 \text{ mGy/min}} = \frac{32 \text{ Gy}}{16 \times 10^{-3} \text{ Gy/min}}$$

$$= 2000 \text{ min} \approx 33.3 \text{ hrs} \approx 1.4 \text{ days.}$$

Q46]  ${}_{27}^{57}\text{Co}$  emits 122 keV  $\gamma$ -rays. L4

energy of each  $\gamma$  is  $E_{\gamma} = 122 \times 10^3 \times 1.6 \times 10^{-19} \text{ J}$

$$A = 1.55 \mu\text{Ci} = 1.55 \times 10^{-6} \times 3.7 \times 10^{10} \text{ Bq} \\ = 57350 \text{ decays/s (Bq)}$$

Radiated energy per second  $E$  is given by

$$E = A E_{\gamma} \quad (\text{in units of J/s})$$

Absorbed energy by person per second is

$$E_{\text{absorbed}} = 0.5 E \quad (\text{J/s})$$

↑ since only 50% of  $\gamma$ -rays interact with the body.

Absorbed energy per day  $E_{\text{tot}} = E_{\text{absorbed}} \times 24 \times 60 \times 60$

$$\therefore E_{\text{tot}} = 0.5 (A E_{\gamma} \frac{\text{J}}{\text{s}}) (24 \times 60 \times 60 \frac{\text{s}}{\text{day}})$$

$$= 0.5 A E_{\gamma} \times 24 \times 3600 \frac{\text{J}}{\text{day}} =$$

$$\Rightarrow AD = \frac{E_{\text{tot}}}{m} = \frac{E_{\text{tot}}}{65} \approx \frac{4.836 \times 10^{-5} \text{ J/day}}{65 \text{ kg}}$$

$$\approx \frac{7.44}{10^7} \left( \frac{\text{J}}{\text{kg}} \right) \times \frac{1}{\text{day}} = 7.44 \times 10^{-7} \frac{\text{Gy}}{\text{day}}$$